

Nanomedicine and Nanoscale Delivery IV

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INTEGRATING
Delivery Science
ACROSS DISCIPLINES



Bottom-up templating of drug nanoparticles in core-shell hydrogel particles for versatile oral drug delivery

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Contact: lattia@mit.edu, pdoyle@mit.edu
Website: doylegroup.mit.edu/



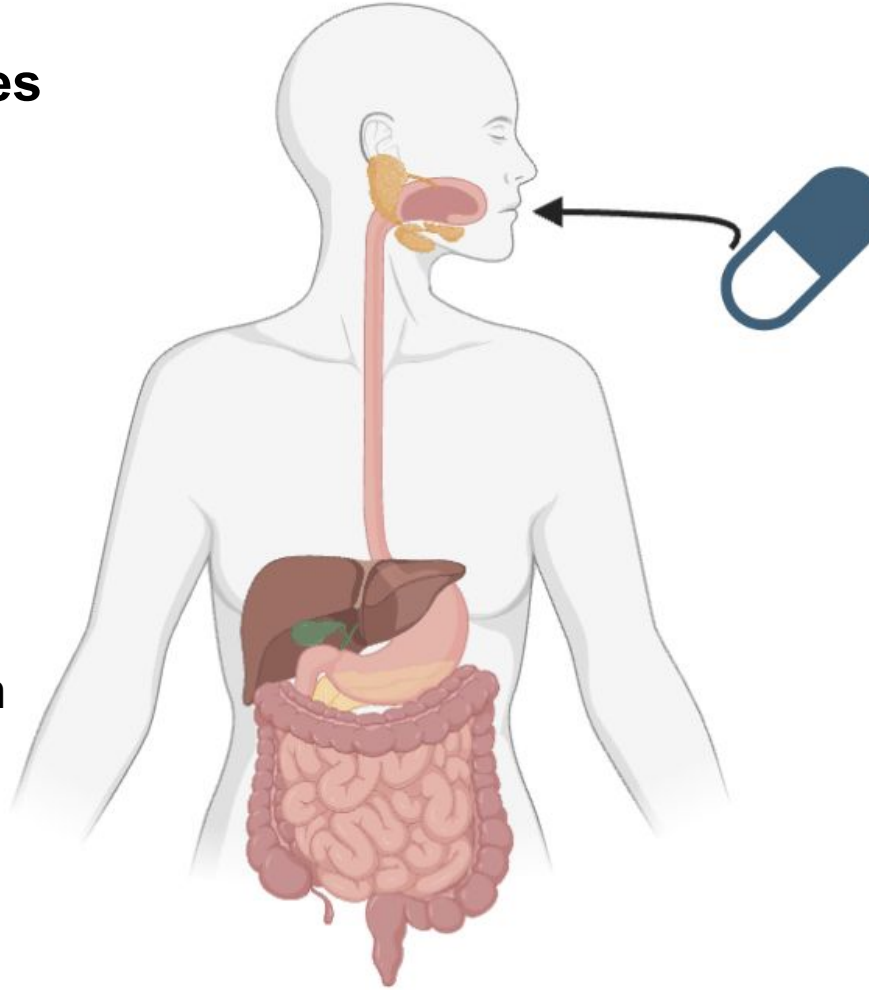
We address two challenges in oral formulation

Generating drug nanoparticles

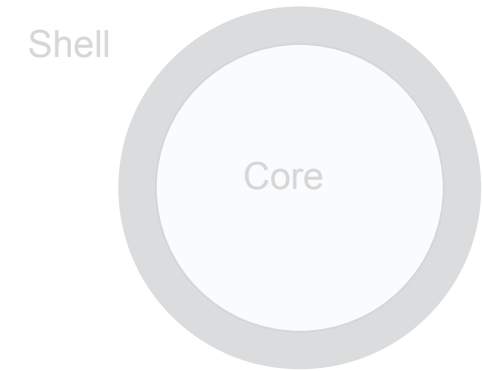


$$O(L) \sim 0.1 - 1 \mu m$$

- ✓ Improve drug dissolution
- ✓ Control crystal size distribution



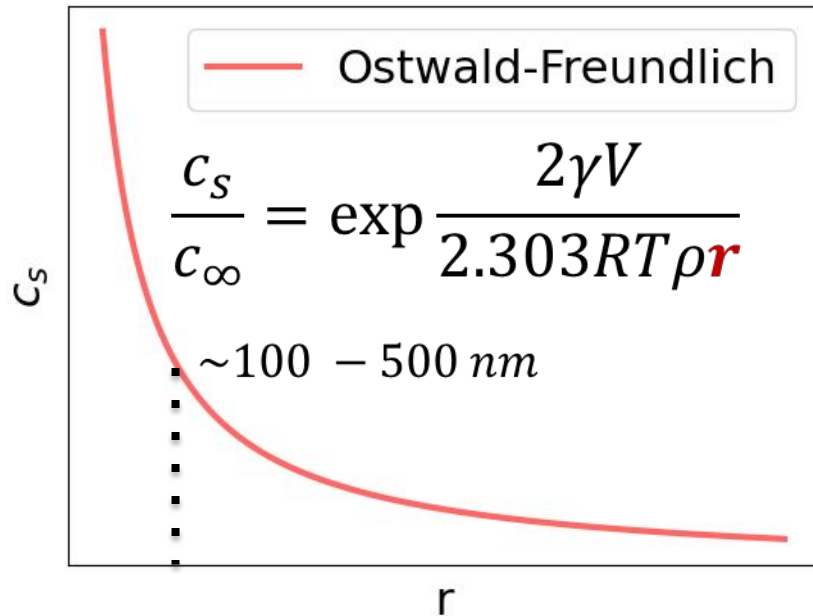
Manufacturing core-shell carriers



- ✓ Control drug release
- ✓ Structure drugs in distinct layers

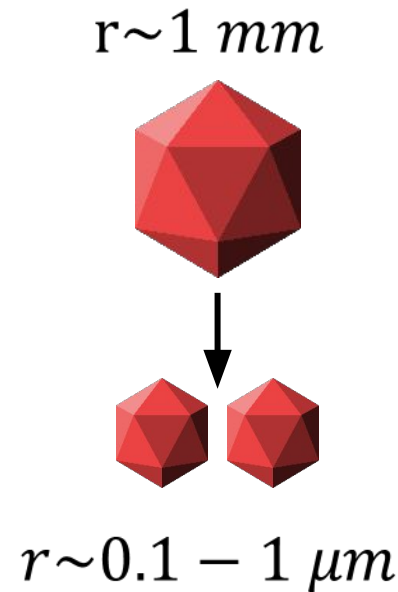
Nanosizing improves the oral bioavailability of hydrophobic APIs

Thermodynamics



Nanosized API particles have increased **equilibrium solubility**

Nanosizing API



Kinetics

Noyes-Whitney

$$\frac{dc}{dt} = \frac{D \times \mathbf{A}(c_s - c_x)}{h}$$

Nanosized API particles have faster **dissolution kinetics**

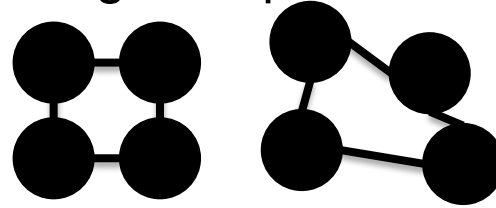
1. Padhye, *J Drug Deliv Sci Technol* (2021)

Conventional 'top-down' nanosizing is resource-intensive and difficult to control

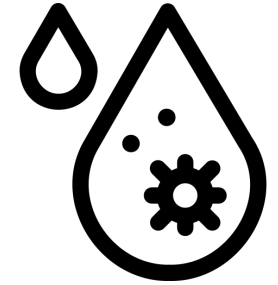
Energy efficiency $< 5\%$



Drug amorphization



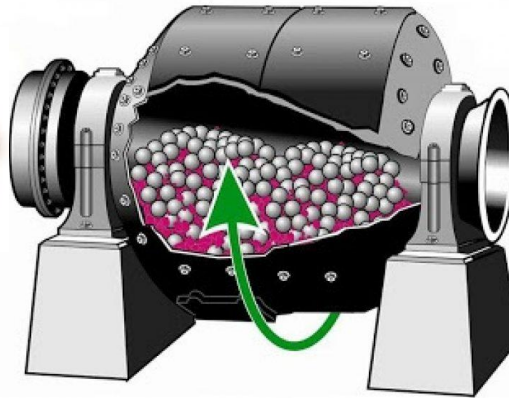
Contaminations from milling beads



$r \sim 1\text{ mm}$



$r \sim 0.1 - 1\text{ }\mu\text{m}$

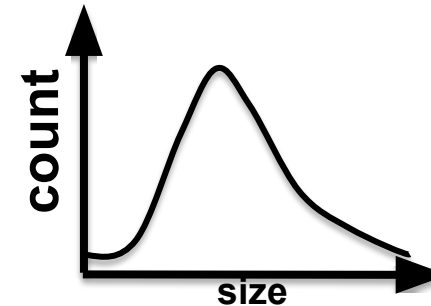


Media Mill

Long residence times
 $\tau \sim 1 - 5\text{ days}$

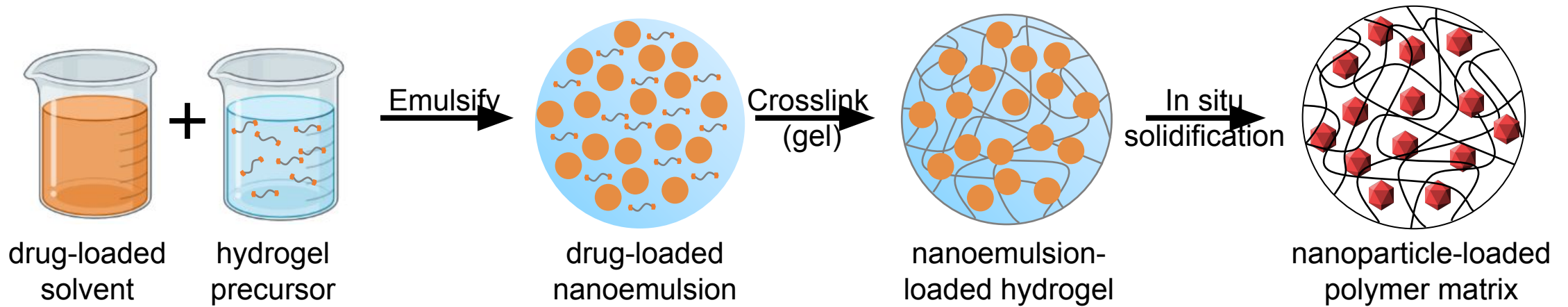


Poor control over
crystal size distribution



1. Schenck. *Mol. Pharm* (2020)
2. Hansuld. *Int. J. Pharm* (2014)

Nanoemulsions template drug nanoparticles from the 'bottom-up'



Eral. *Chem. Mater.* (2014)
Eral. *Cryst. Growth Des.* (2014)
Badrudodoza. *Adv. Healthc. Mater.* (2016)
Badrudodoza. *Adv. Ther.* (2018)

Domenech. *Chem. Mater.* (2020)
Chen. *Adv. Mater.* (2021)
Chen. *Chem. Mater.* (2022)
Attia. *Adv. Healthc. Mater.* (2023)

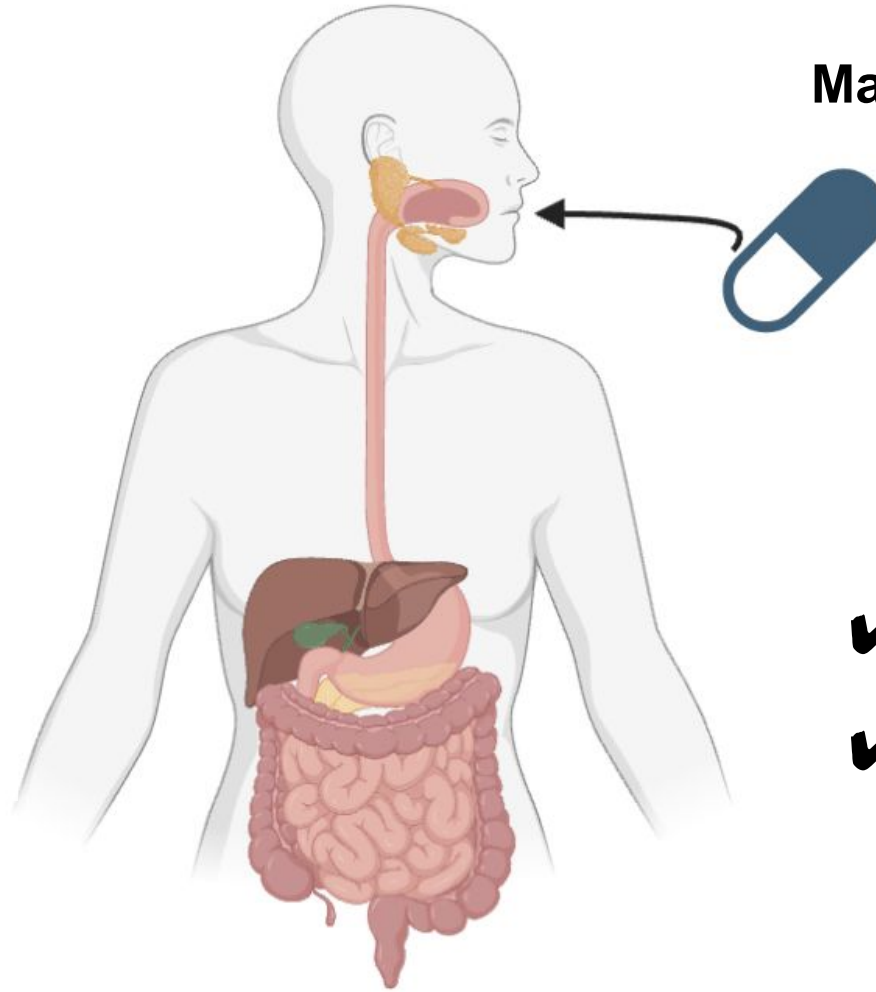
We address two challenges in oral formulation

Generating drug nanoparticles



$$O(L) \sim 0.1 - 1 \mu m$$

- ✓ Improve drug dissolution
- ✓ Control crystal size distribution



Manufacturing core-shell carriers

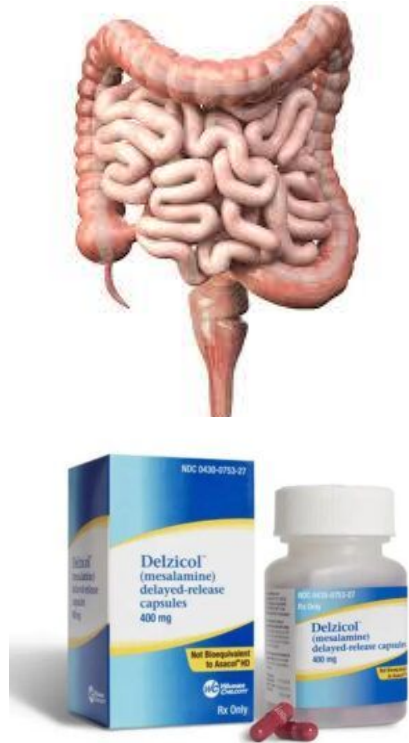
Shell

Core

- ✓ Control drug release
- ✓ Structure drugs in distinct layers

Various therapeutic applications require core-shell carriers

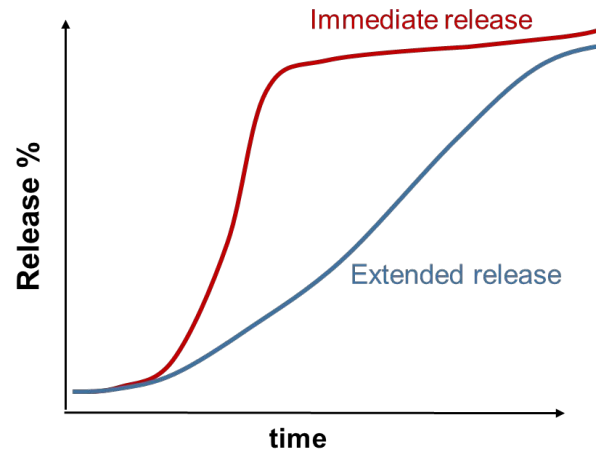
Intestinal and colonic targeting



Delzicol

McCoubrey. *J. Control. Release* (2023)

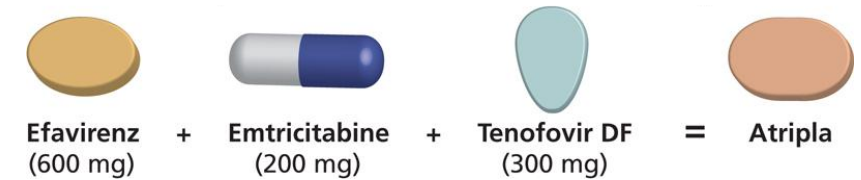
Sustained/Extended release



Carbamazepine

McHugh, *Nature Reviews Drug Discovery* (2023)

Combination drug products



Atripla

Rabbani. *Am. J. Hypertens.* (2008)

Current core-shell carrier manufacturing platforms are limited

Platform

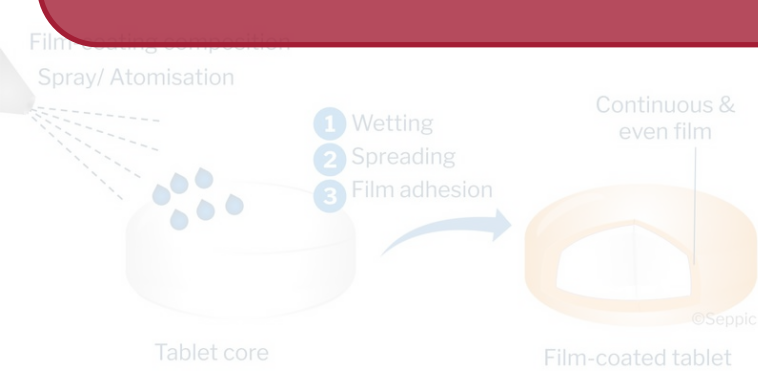
Additional unit operation

Organic solvent

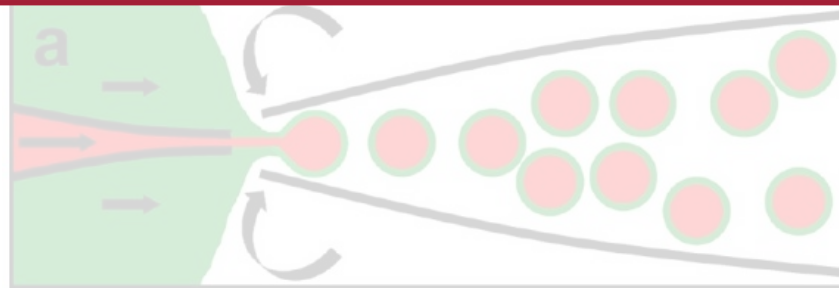
Structure drugs in
distinct layers

High processing
temperature

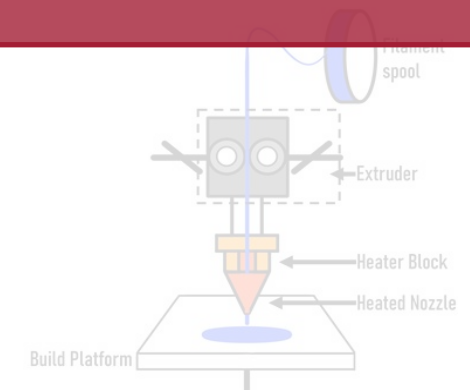
1. Manufacture core-shell particles with distinct polymeric layers
2. Template hydrophobic drug nanoparticles in each layer
3. Control release kinetics through formulation parameters



1. Film coating. SEPPIC

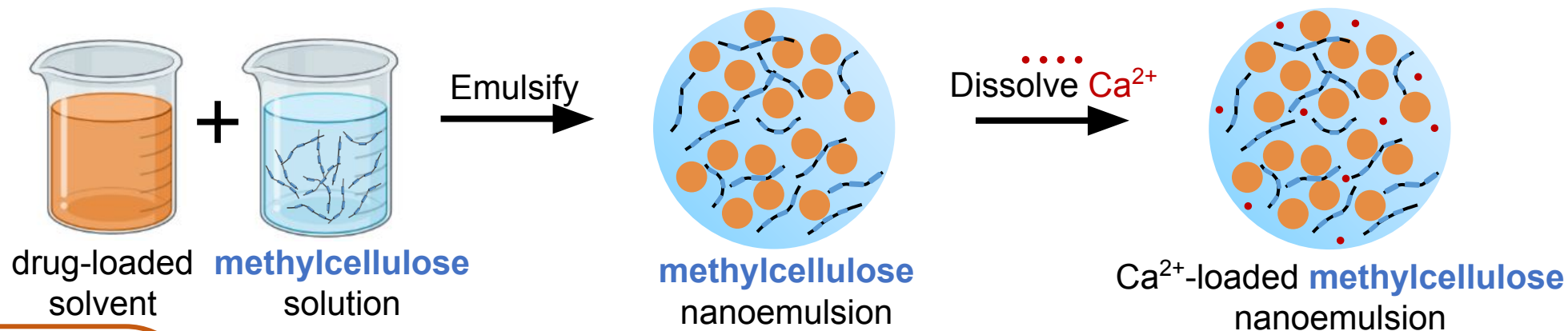


2. Windbergs. JACS. (2013)

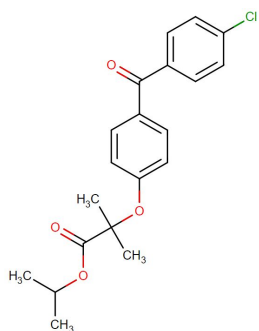


3. Parulski. Adv. Drug Deliv. Rev. (2021)

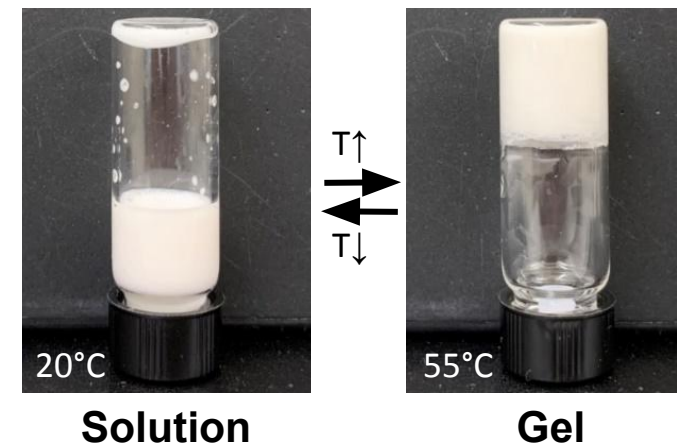
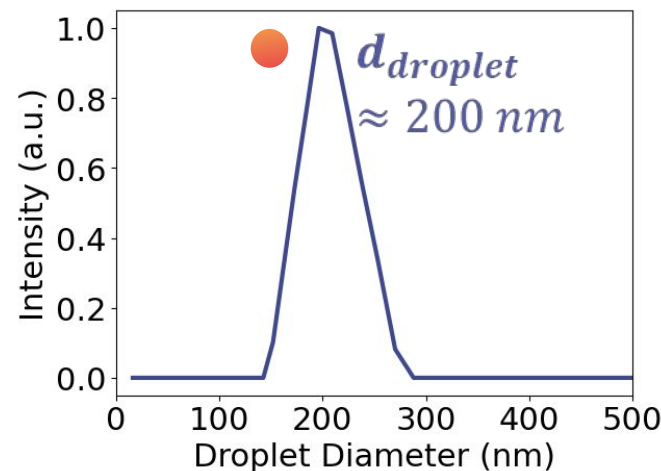
We combine thermogelling and ion-gelling capabilities



Fenofibrate (FEN)¹
 $0.3 \mu\text{g mL}^{-1}$ water

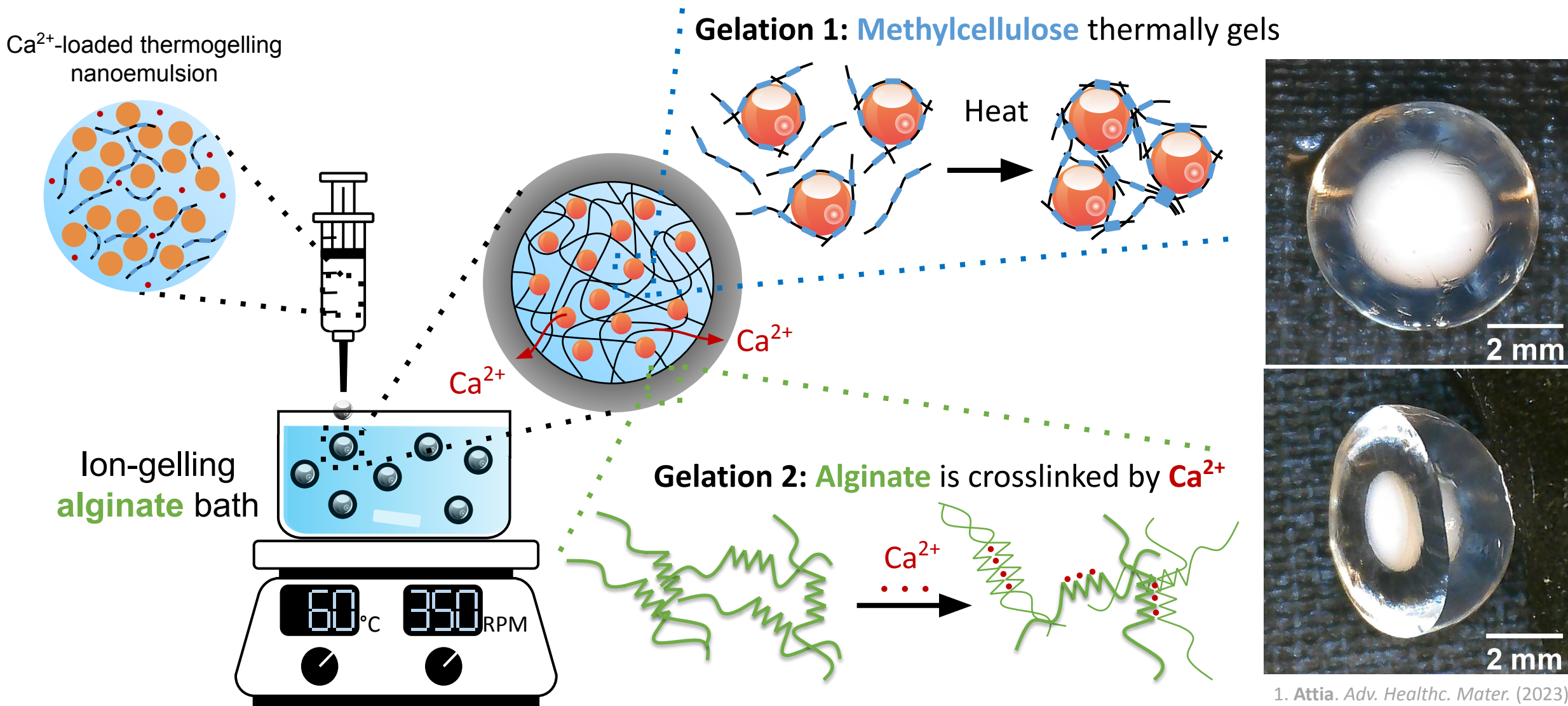


- ✓ Low solubility
- ✓ High permeability



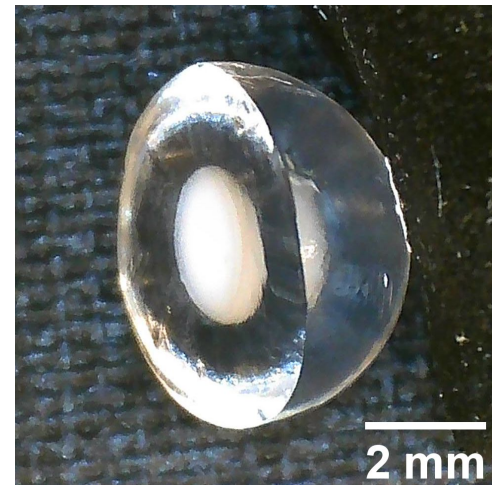
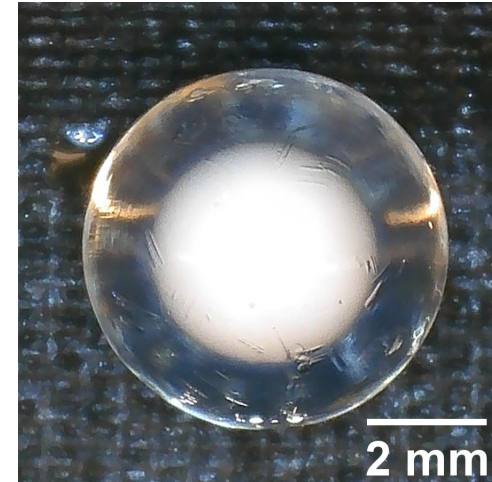
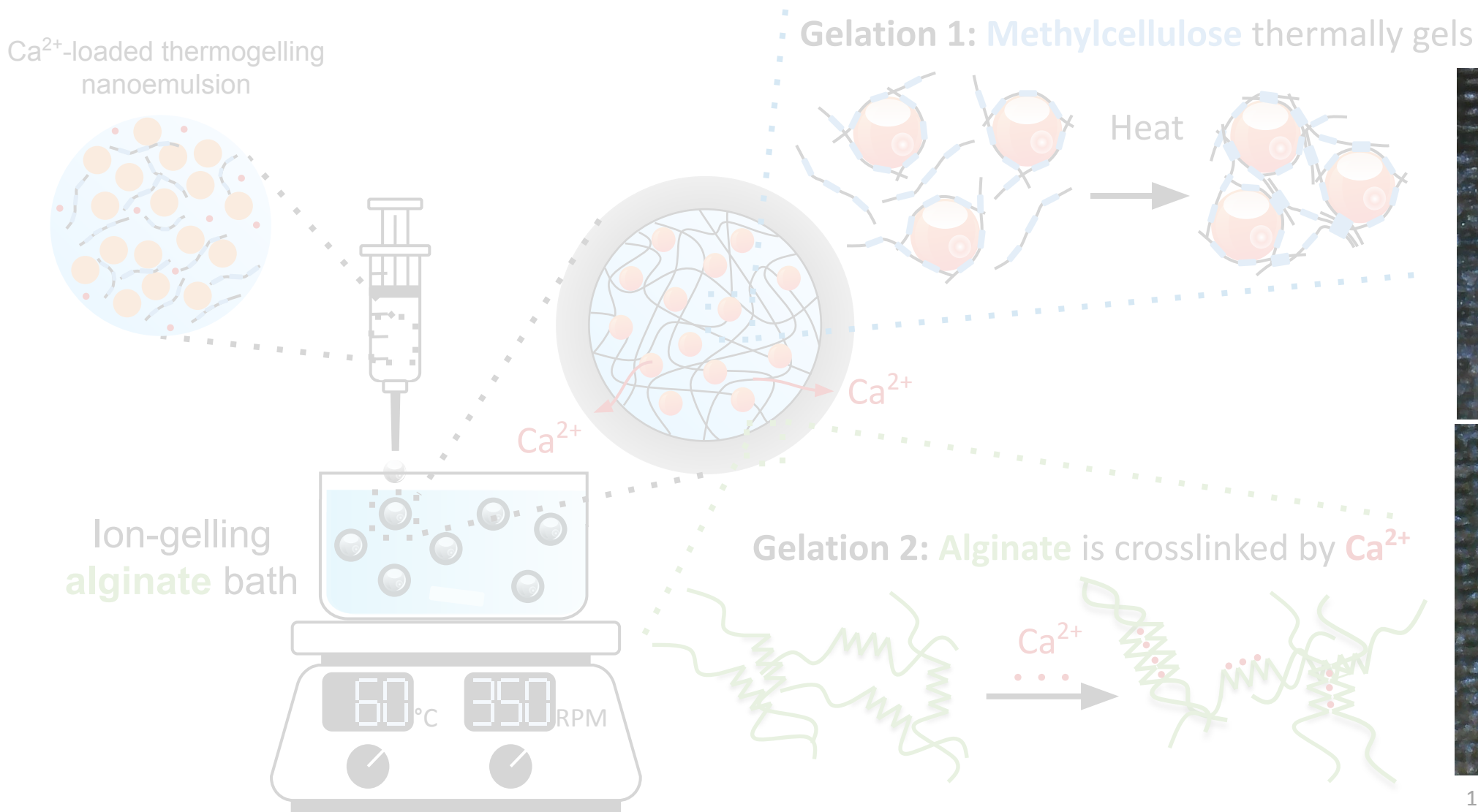
1. Tipduangta. *Cryst. Growth Des.* (2015)

We exploit these orthogonal gelling mechanisms



1. Attia. *Adv. Healthc. Mater.* (2023)

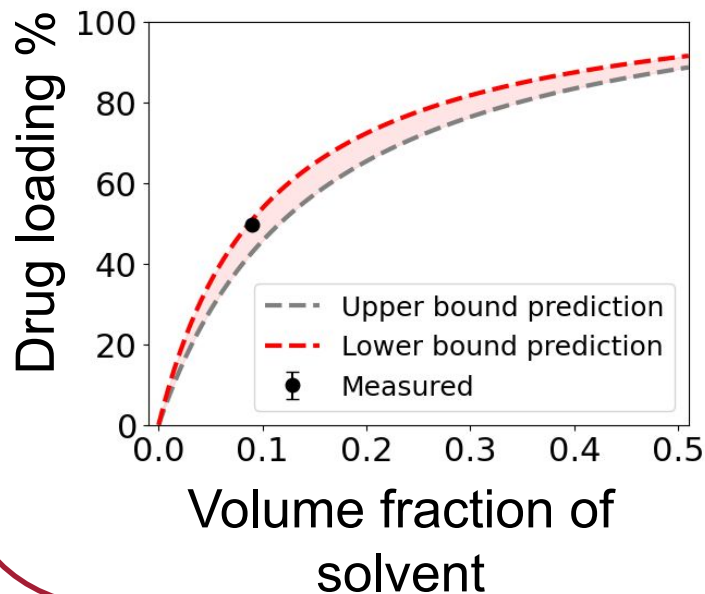
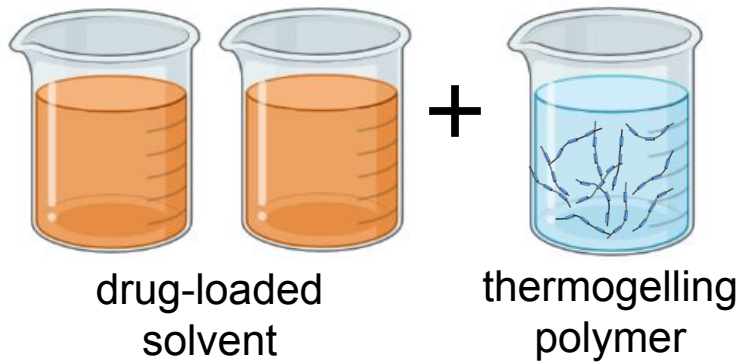
We exploit these orthogonal gelling mechanisms



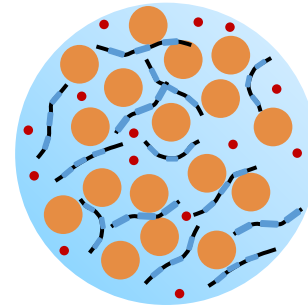
1. Attia. *Adv. Healthc. Mater.* (2023)

Formulation parameters control core-shell particle properties

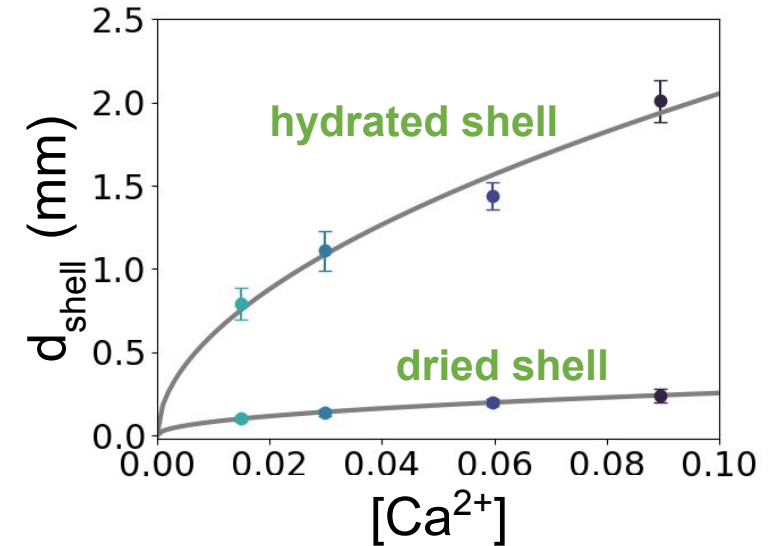
Volume fraction of solvent droplets controls drug loading



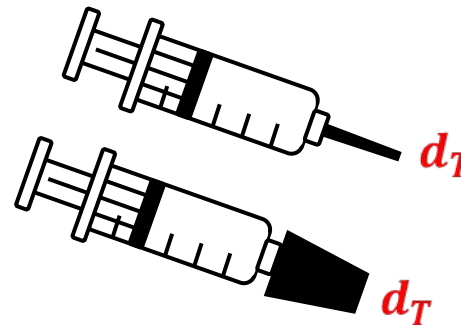
Salt concentration controls the **alginate** shell thickness



Ca²⁺-loaded thermogelling nanoemulsion



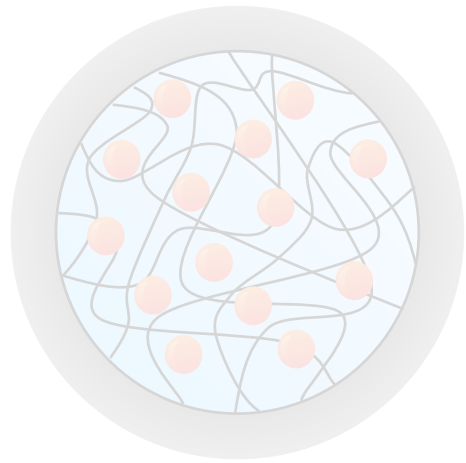
Dispensing tip size controls **methylcellulose** core radius



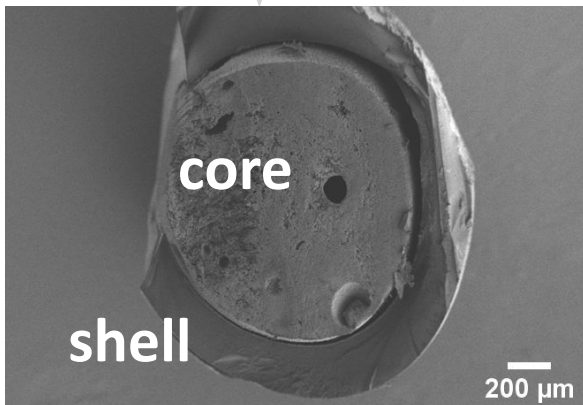
Tate's Design Law

$$r_{\text{core}} \propto \left(\frac{d_T \gamma_{NE}}{\rho_{NE} g} \right)^{\frac{1}{3}}$$

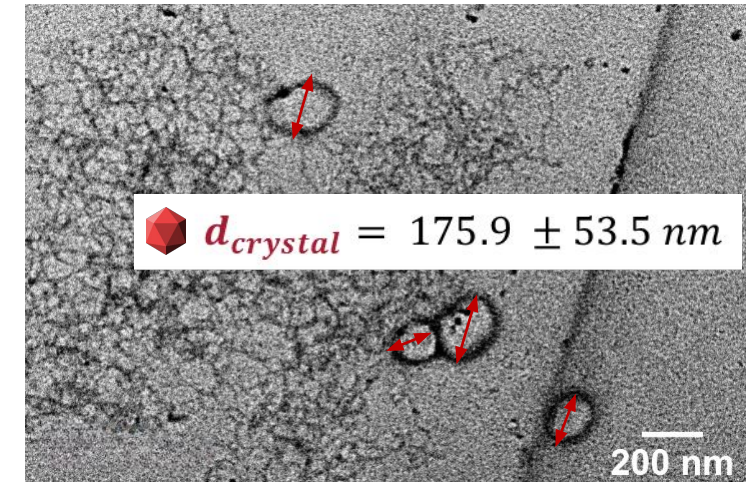
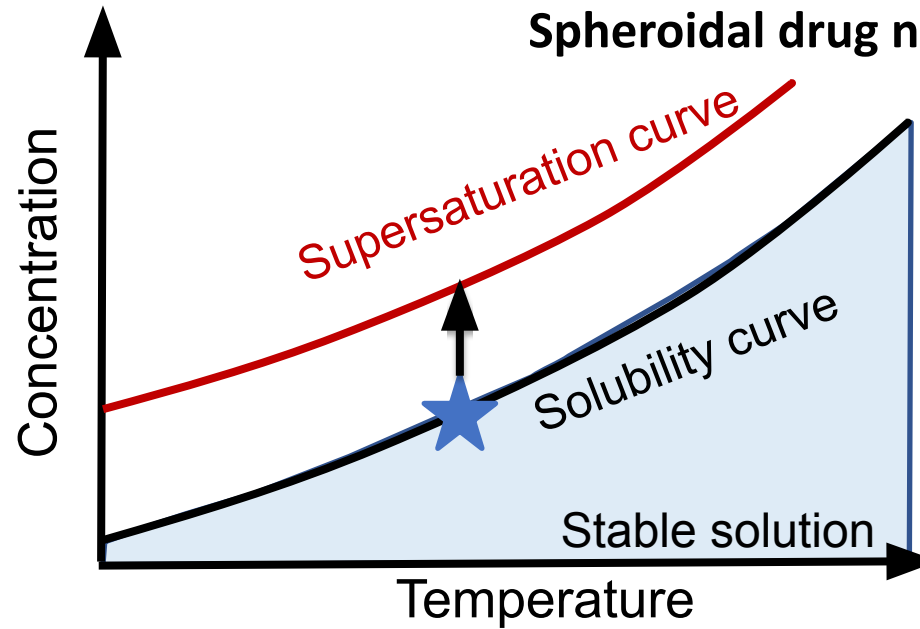
In situ crystallization templates drug nanoparticles



70 °C
Solvent
extraction



Spheroidal drug nanocrystals are templated

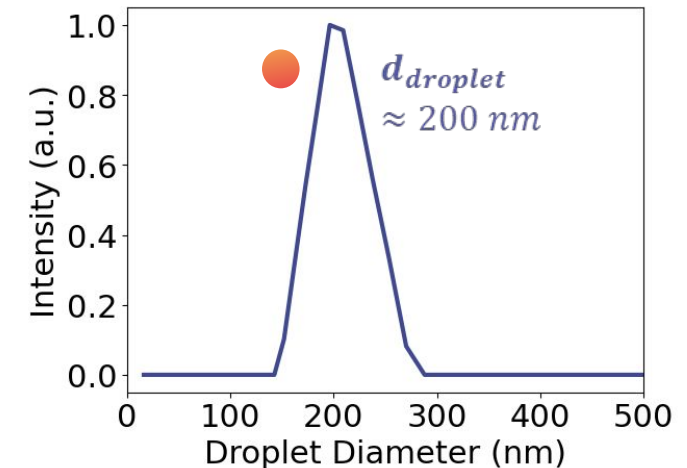


Nanodroplet size distribution controls crystal size

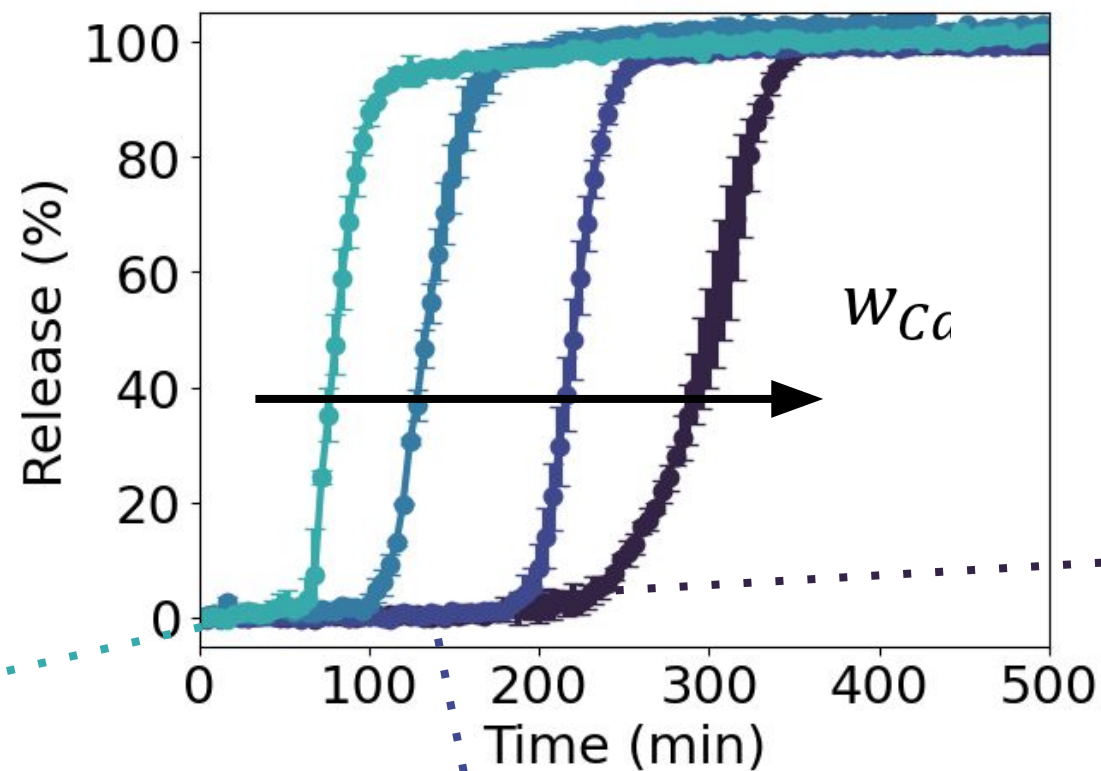
$$m_{drug,crystal} = m_{drug,droplet}$$

$$\frac{4}{3}\pi \left(\frac{\hat{d}_{crystal}}{2} \right)^3 \rho_{crystal} = \frac{4}{3}\pi \left(\frac{d_{droplet}}{2} \right)^3 C_{drug}$$

$$\hat{d}_{crystal} = \left(\frac{C_{drug}}{\rho_{crystal}} \right)^{\frac{1}{3}} d_{droplet} = 150 \text{ nm}$$

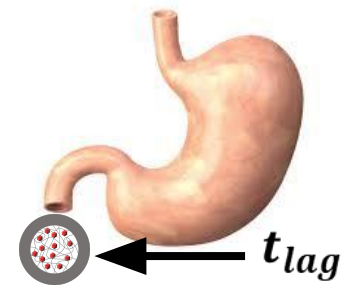


Increasing shell thickness accesses site-specific delayed release profiles

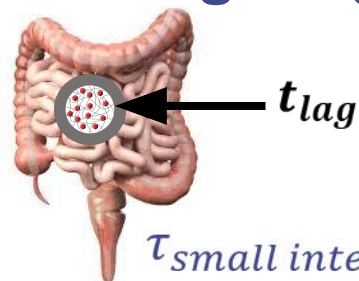


Enteric coating

$\tau_{stomach} \sim 1 - 2 \text{ hours}$



Intestinal targeting¹



$\tau_{small \text{ intestine}} \sim 3 - 5 \text{ hours}$

Colonic targeting²

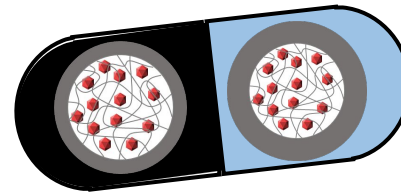
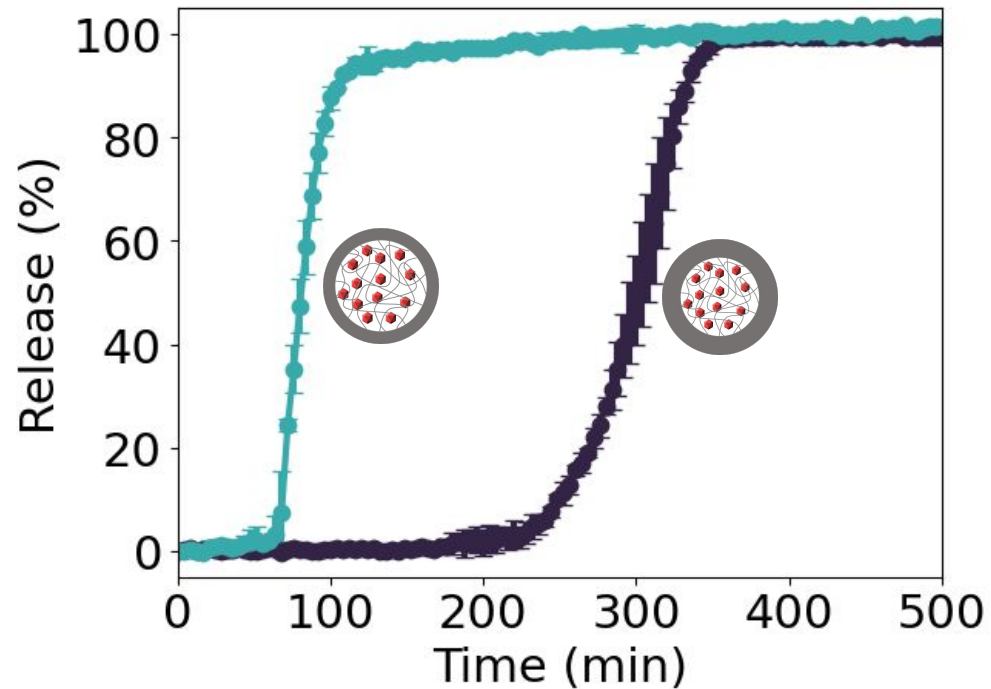
$\tau_{large \text{ intestine}} \sim 4 + \text{hours}$



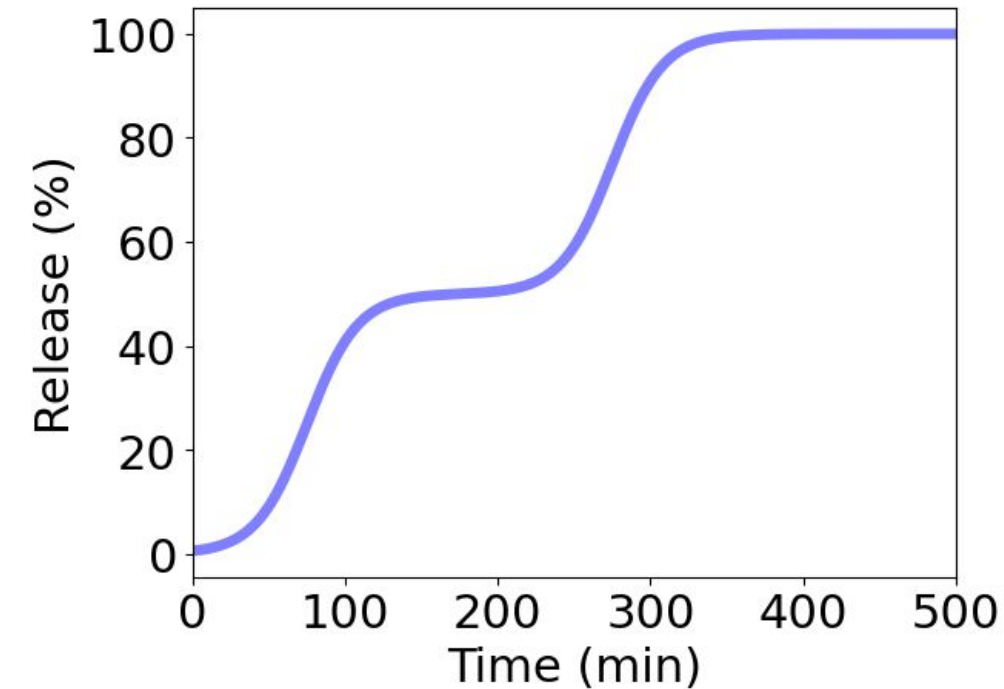
1. Hua, S., *Front. Pharmacol.* (2020)
2. Amidon S., *AAPS PharmSciTech* (2015)

Combining particles with different lag times can sustain drug release

In vitro drug release results

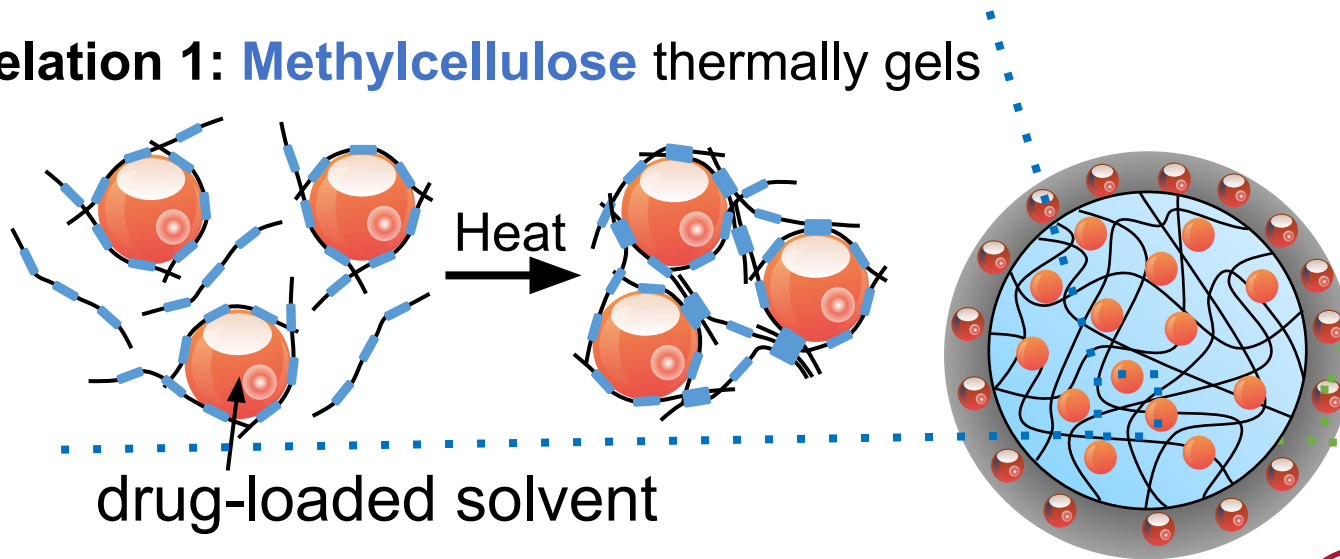


Simulated sustained release

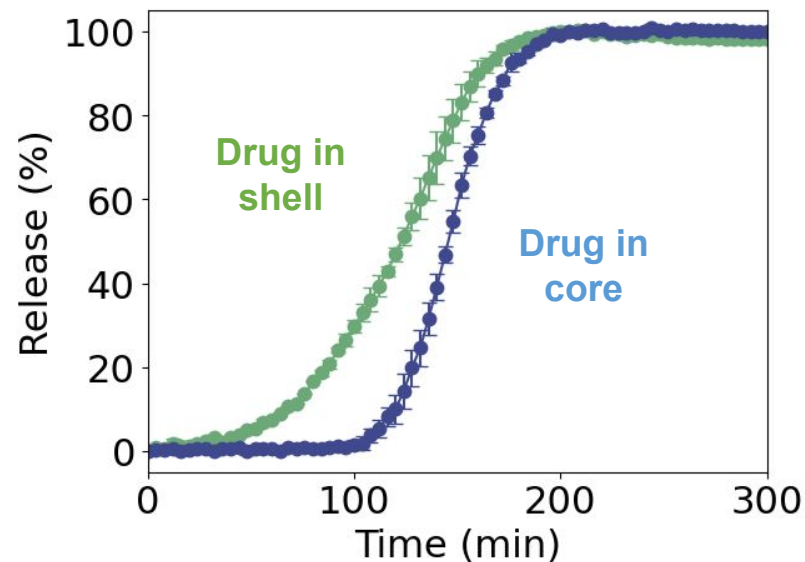
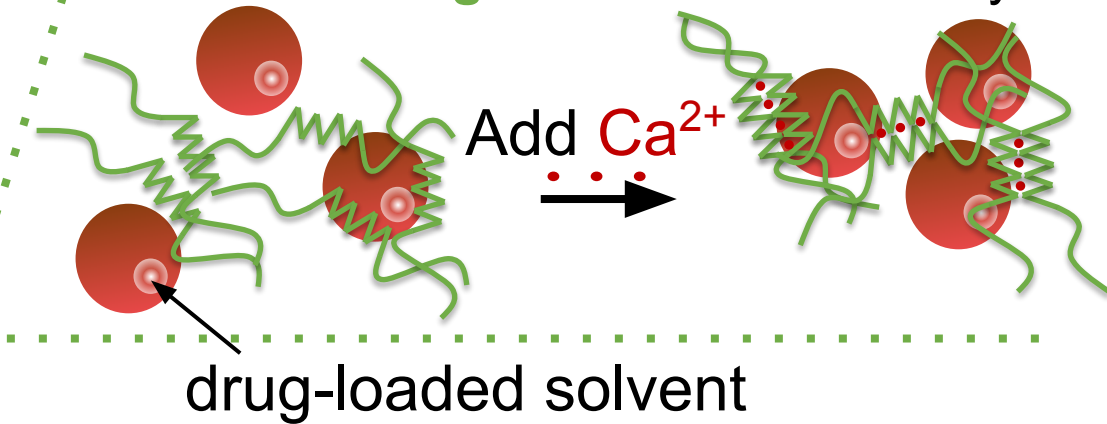


Our approach can structure drug nanoparticles in each gel layer

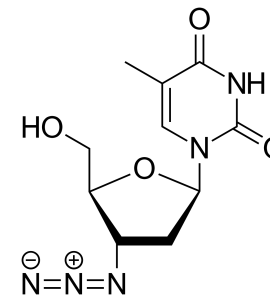
Gelation 1: **Methylcellulose** thermally gels



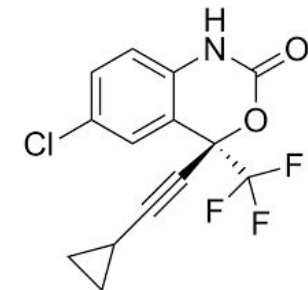
Gelation 2: **Alginate** is crosslinked by Ca^{2+}



Zidovudine
(anti-retroviral)



Efavirenz
(anti-retroviral)



Example drug combination¹

1. Gomes. *J. Braz. Chem. Soc.* (2013)

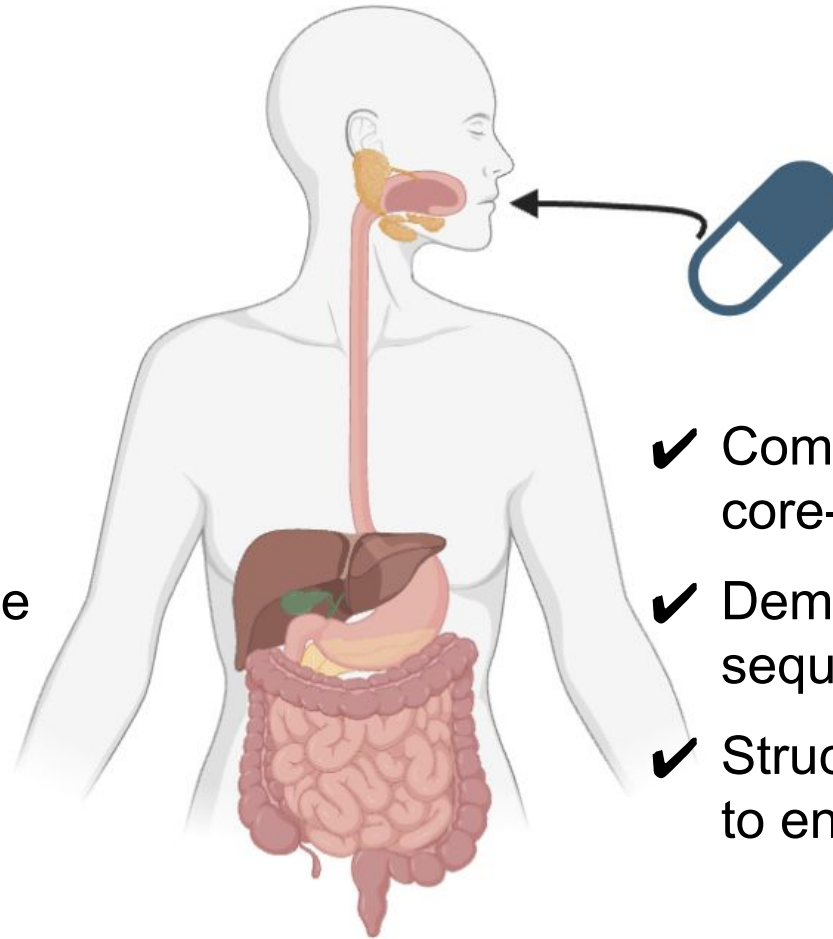
Summary

Generating drug nanoparticles

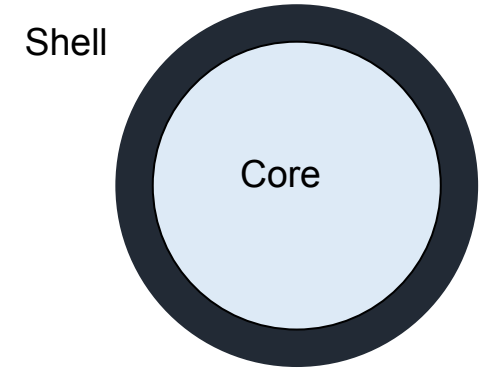


$$O(L) \sim 0.1 - 1 \mu m$$

- ✓ Nanoemulsions template drug nanoparticles inside hydrogels
- ✓ Droplet size controls nanoparticle size
- ✓ Templated nanoparticles have improved drug dissolution



Manufacturing core-shell carriers



- ✓ Combined orthogonal gelations to build core-shell hydrogels in 1 step
- ✓ Demonstrated delayed, sustained, and sequential drug release
- ✓ Structured different drugs in each layer to enable new combination products

Thank you! Questions?

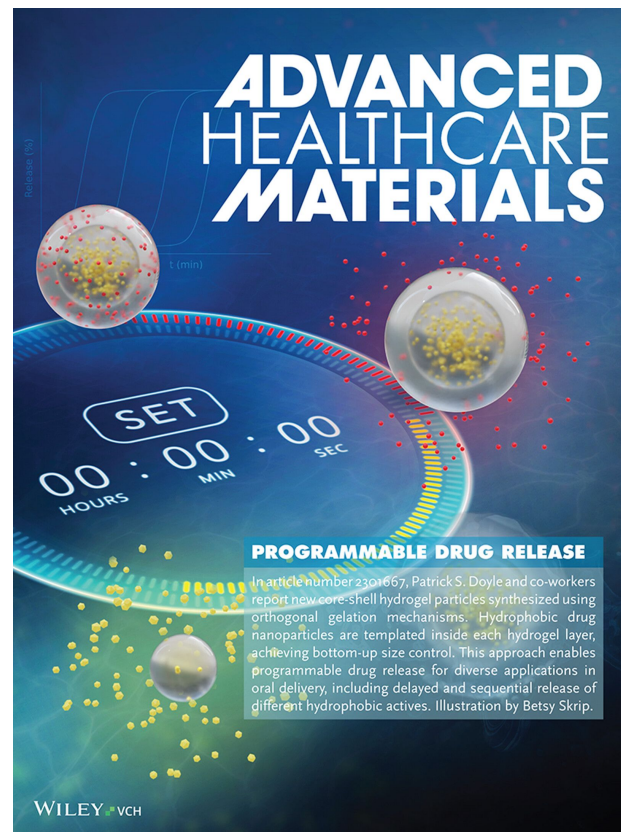
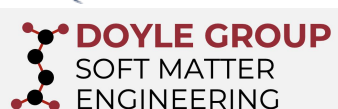
Acknowledgments

Doyle Group

- Prof. Patrick Doyle
- Liang-Hsun Chen
- Talia Zheng
- Swati Shikha
- Amir Erfani

Characterization


- Yong Zhang
- Nicole Bohn



QR code
to paper



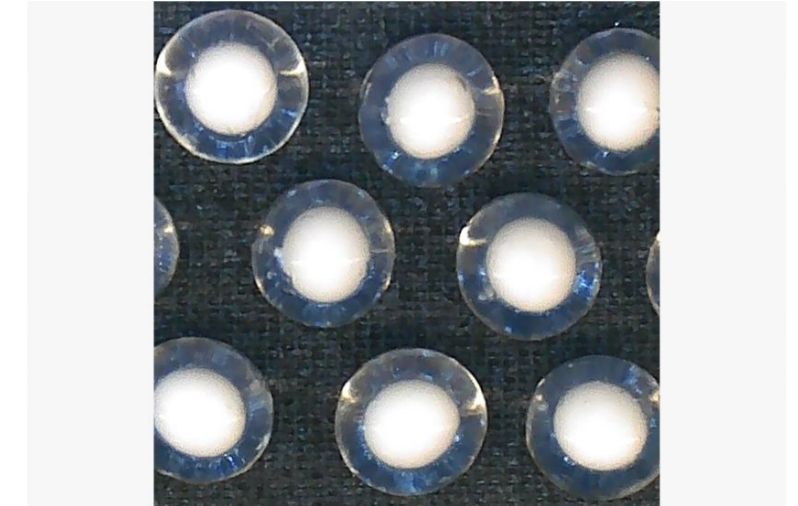
MIT News
ON CAMPUS AND AROUND THE WORLD

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A new way to deliver drugs more efficiently

Core-shell structures made of hydrogel could enable more efficient uptake in the body.

Department of Chemical Engineering
November 28, 2023

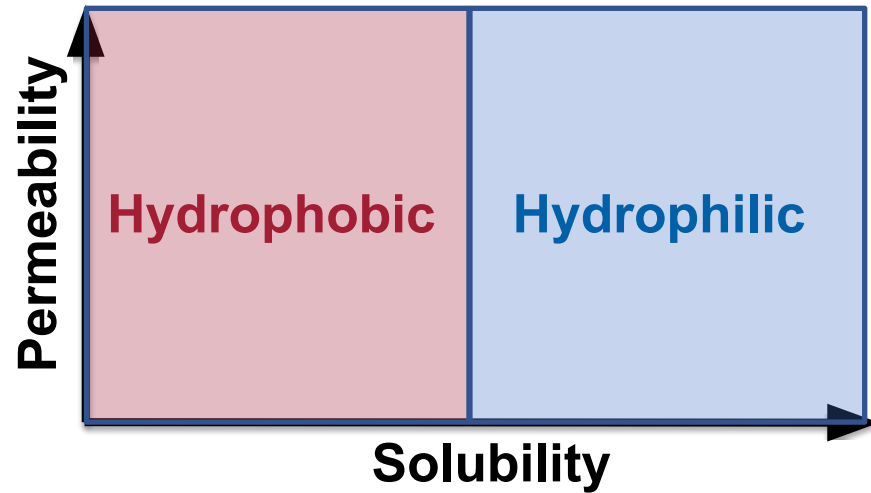


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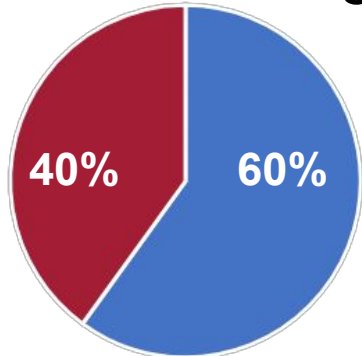


Hydrophobicity limits the oral delivery of small molecules

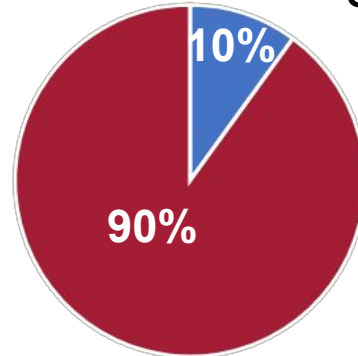
Most drug molecules are hydrophobic



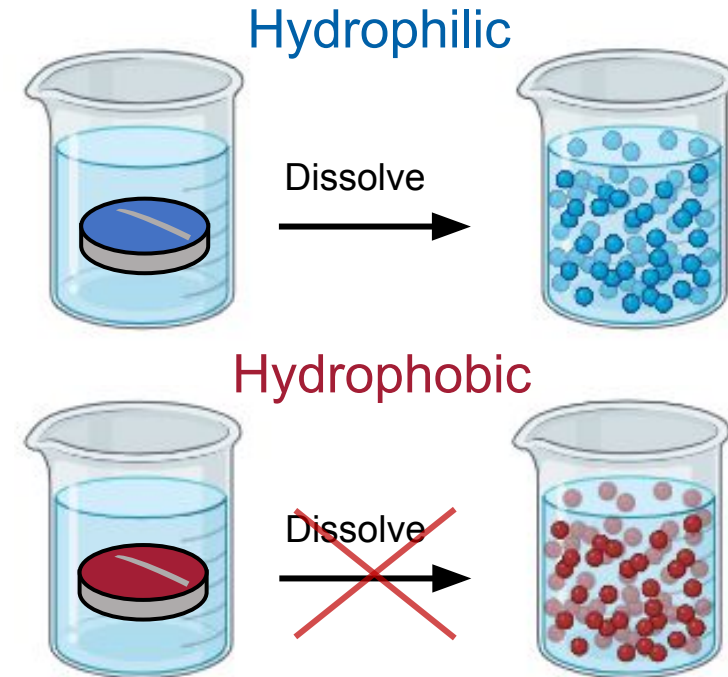
Marketed drugs



Candidate drugs¹



Hydrophobic drugs have poor dissolution



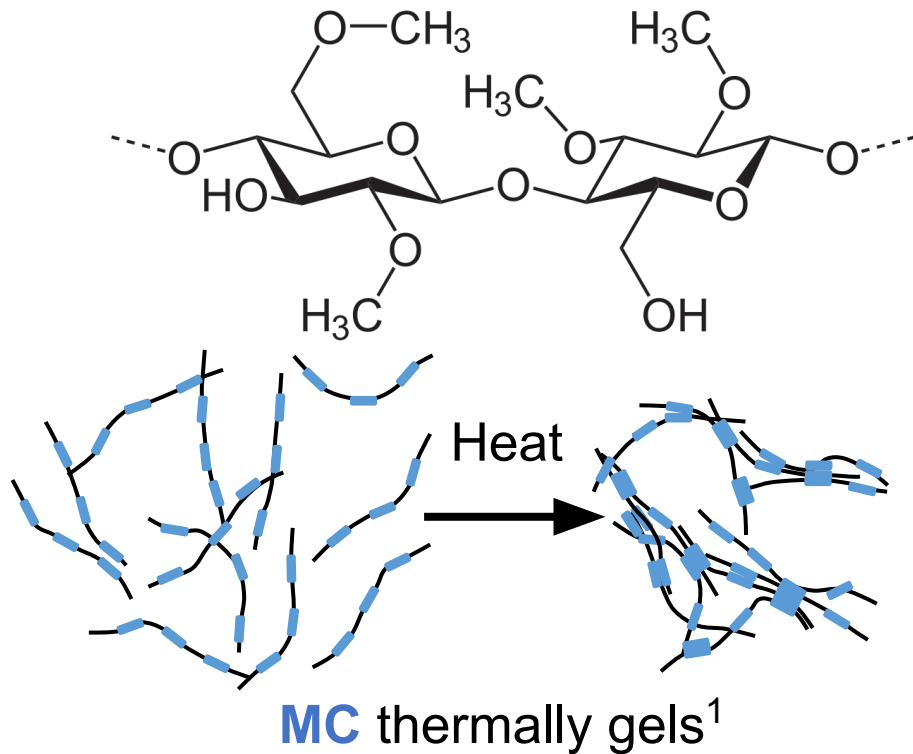
Causes 10-15 % of clinical trial failures²

1. Alqahtani. *Front. Pharmacol.* (2021)

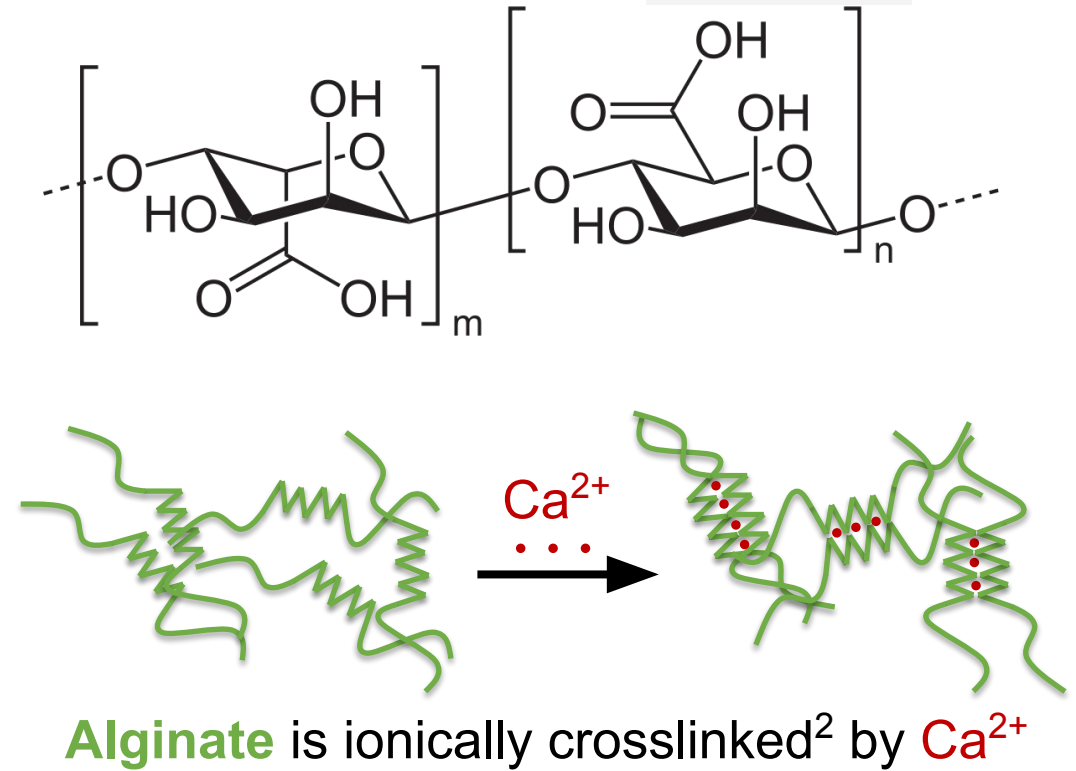
2. Brouwers. *J. Pharm. Sci.* (2009).

We use nature-derived polymers as model gelators

Methylcellulose (MC)



Alginate

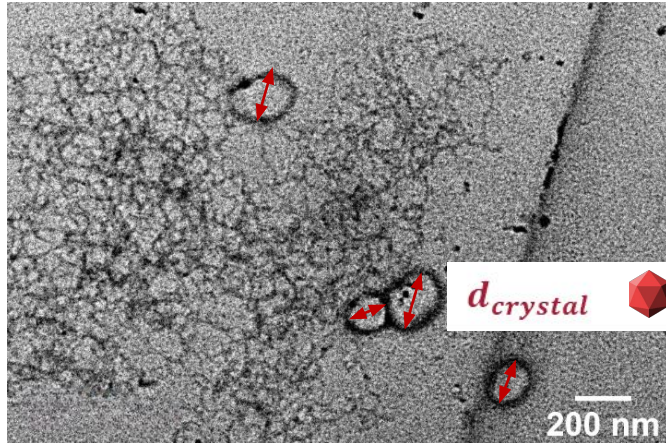


1. Chen. *Adv. Mater.* (2021)

2. Velings. *Polym. Gels Networks* (1995)

Nanodroplet size distribution controls nanoparticle size

Crystal size distribution



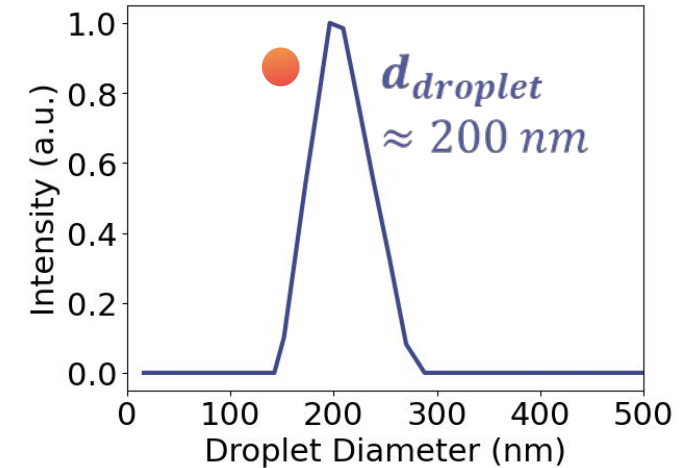
$$d_{crystal} = 175.9 \pm 53.5 \text{ nm}$$

$$m_{drug,crystal} = m_{drug,droplet}$$

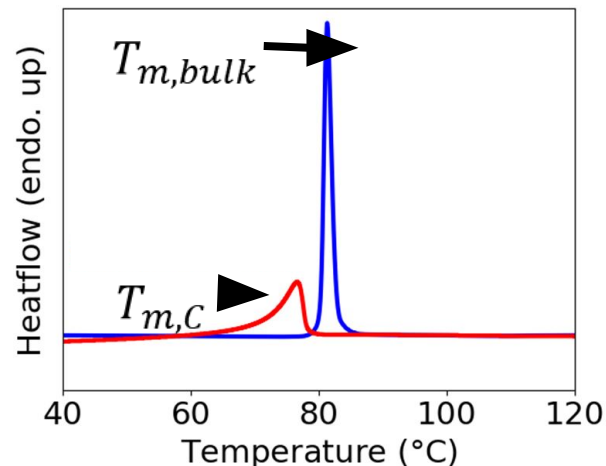
$$\frac{4}{3}\pi \left(\frac{\hat{d}_{crystal}}{2} \right)^3 \rho_{crystal} = \frac{4}{3}\pi \left(\frac{d_{droplet}}{2} \right)^3 c_{drug}$$

$$\hat{d}_{crystal} = \left(\frac{c_{drug}}{\rho_{crystal}} \right)^{\frac{1}{3}} d_{droplet} = 150 \text{ nm}$$

Nanodroplet size distribution



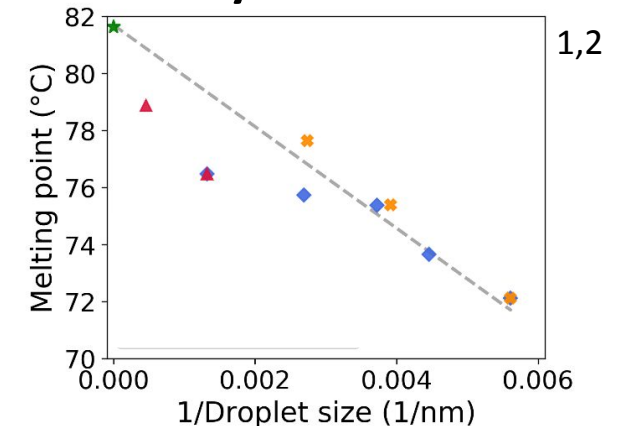
Melting point depression validates the relationship between droplet size and crystal size



Gibbs-Thompson Equation

$$T_{m,C} \sim \left(1 - \frac{K}{d_{crystal}} \right) T_{m,bulk}$$

$$T_{m,C} \sim \left(1 - \frac{K}{d_{droplet}} \right) T_{m,bulk}$$



1. Chen. *Adv. Mater.* (2021)

2. Dwyer. *CrystEngComm.* (2015)